

# Estimation of Refrigerant-Oil Mixture Viscosities for Alternative Refrigerants Using Solubility Data

Chris Seeton

Seeton C&P

University of Illinois at Urbana-Champaign

Air Conditioning and Refrigeration Center

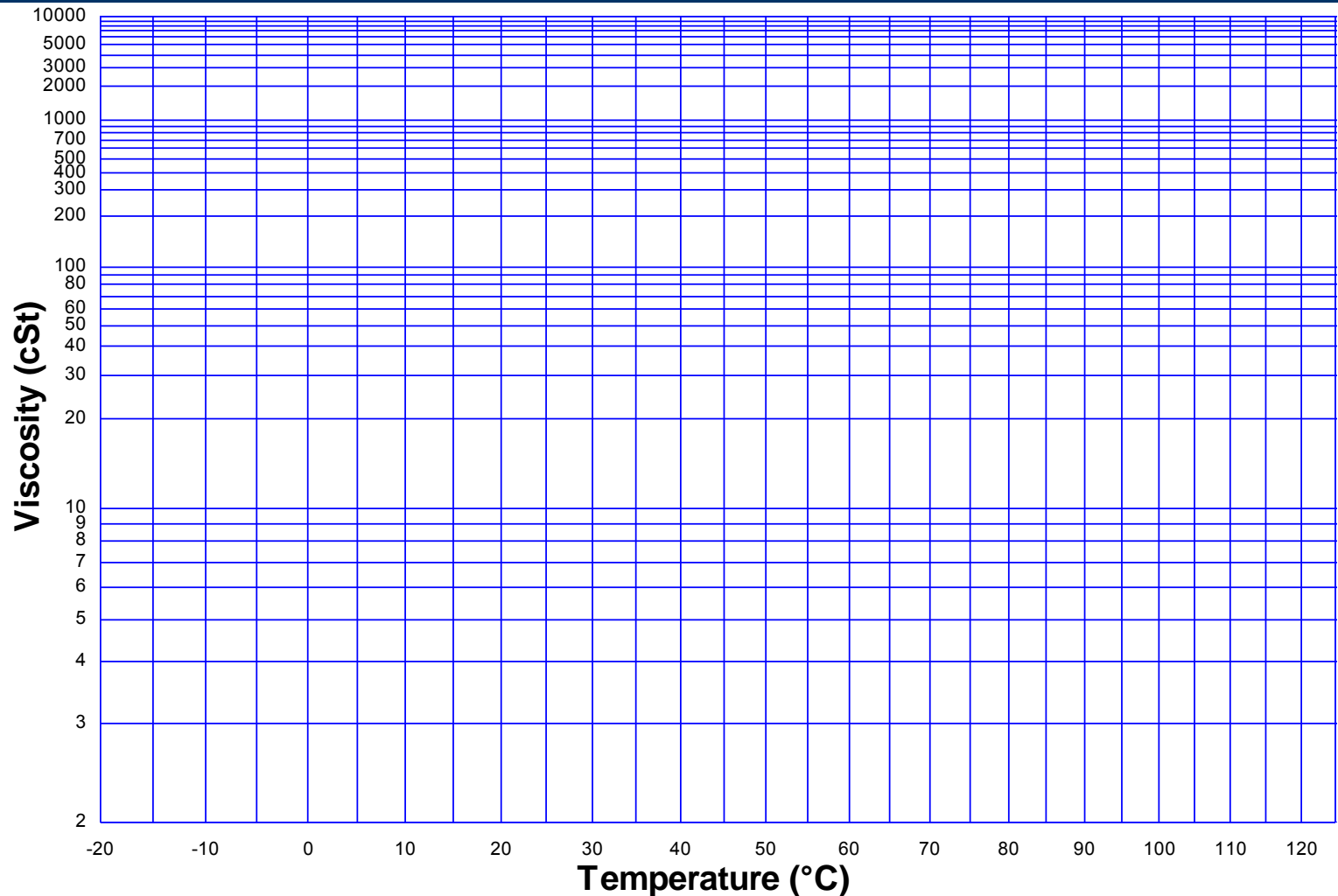
[seeton@uiuc.edu](mailto:seeton@uiuc.edu)

# Why Estimate/Measure Mixture Viscosity?

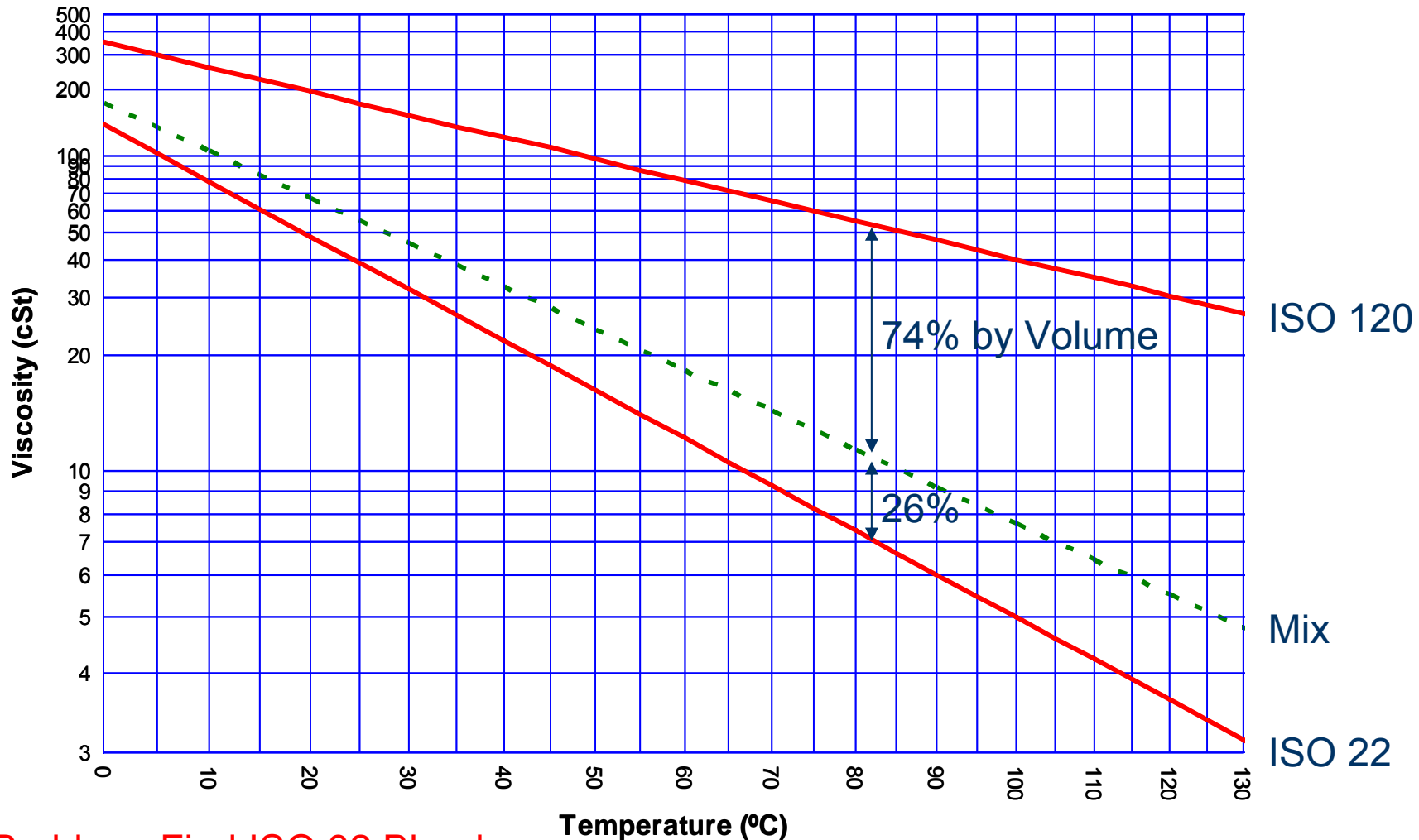
- Refrigerant drastically cuts the lubricant viscosity
  - **Primarily: Compressor Lubrication**
  - Secondary:
    - Lubricating auxiliary valves and seals
    - Sealing microscopic leaks
  - Heat Exchanger / Piping Design Considerations
    - Heat transfer enhancement or fouling
      - Heat Exchanger length -  $Nu \sim f(Pr, Re, \dots)$
    - Increased Pressure drop
      - Heat Exchanger circuiting or tube sizing -  $\Delta P \sim f(Re, \dots)$
    - Oil Management
      - Retention, “Hold-up” & Oil Return

$$Re = \left( \frac{\vec{V}D}{\nu} \right)$$

# Viscosity – Temperature Chart

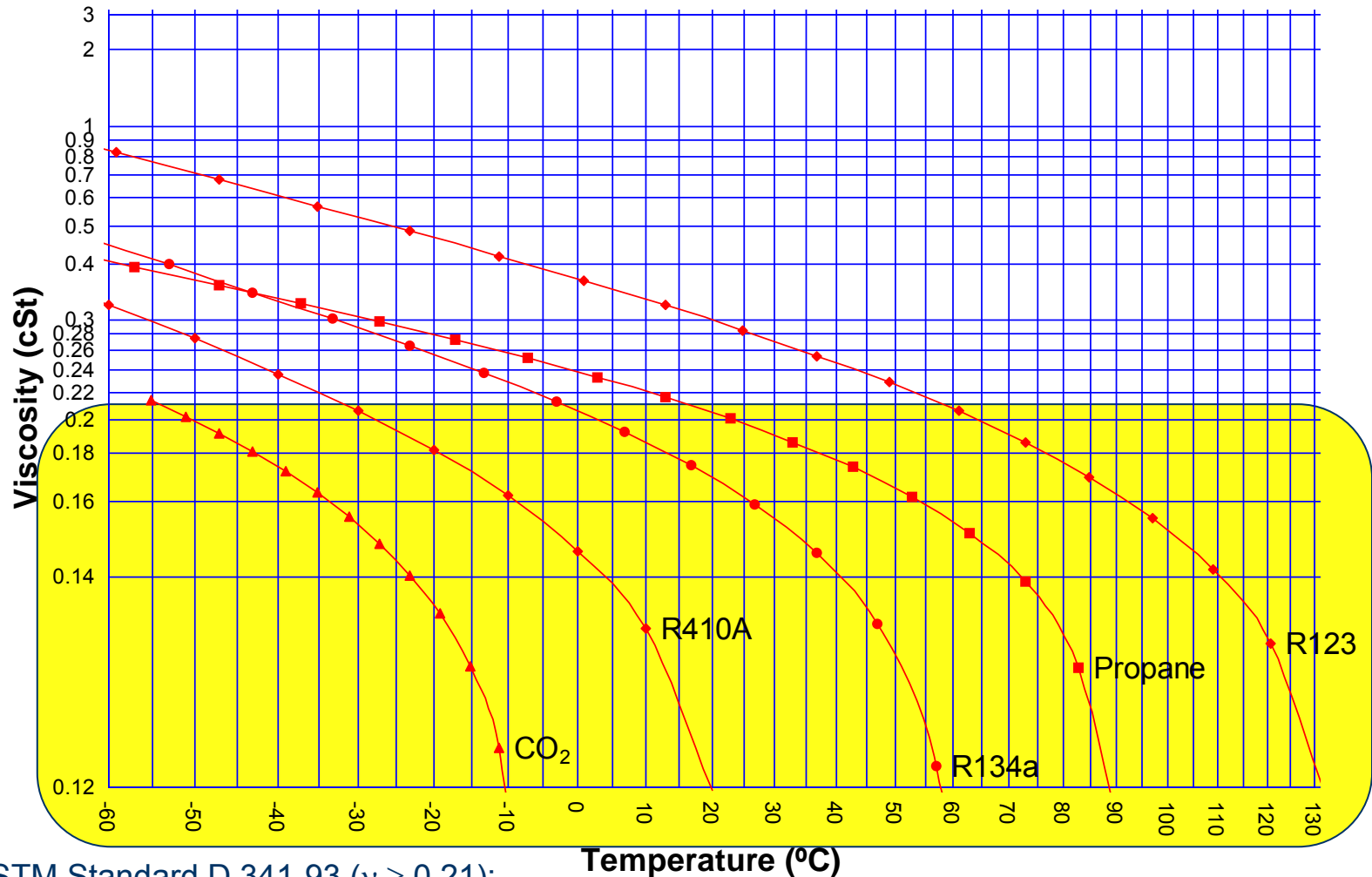


# Historical Lubricant Blending



Problem: Find ISO 32 Blend

# ASTM Chart – Application to Refrigerants



ASTM Standard D 341-93 ( $\nu \geq 0.21$ );  
Manning equation ( $\nu < 0.21$ )

# 2006 Refrigeration Handbook, Chapter 7

- Relationship between temperature and kinematic viscosity (ASTM D341)

$$\text{Log}_{10} \left( \text{Log}_{10} \left( \nu + 0.7 + f_{ASTM}(\nu) \right) \right) = A - B * \text{Log}_{10}(T)$$

$$f_{ASTM}(\nu) = C - D + E - F + G - H$$

$$C = \exp(-1.14883 - 2.65868\nu) \quad ; \quad D = \exp(-0.00381308 - 12.5645\nu)$$

$$E = \exp(5.46491 - 37.6289\nu) \quad ; \quad F = \exp(13.0458 - 74.6851\nu)$$

$$G = \exp(37.4619 - 192.643\nu) \quad ; \quad H = \exp(80.4945 - 400.468\nu)$$

- Undefined below 0.21 cSt!

# Criteria for a new Viscosity-Temperature Chart

It is desired to construct a generalized chart that is able to:

- Cover entire temperature range (cryogenic to high temperatures)
- Cover entire viscosity range (~0.04 cSt to glass transition)
- Linearize fluids that do not exhibit excessive molecular coiling, molecular bonding, or wax precipitation
- Maintain the existing ASTM format for lubricants for viscosities greater than 2 cSt.
- **Provide blending of lubricant and refrigerant mixtures**

# New Scaling Rule

$$\text{Log}_e \left( \text{Log}_e \left( \nu + 0.7 + f_{CS}(\nu) \right) \right) = A - B * \text{Log}_e(T)$$

$$f_{CS}(\nu) = e^{-\nu} K_0(\nu + \psi)$$

- Scaling the LHS to approach -infinity as  $\nu \rightarrow 0$

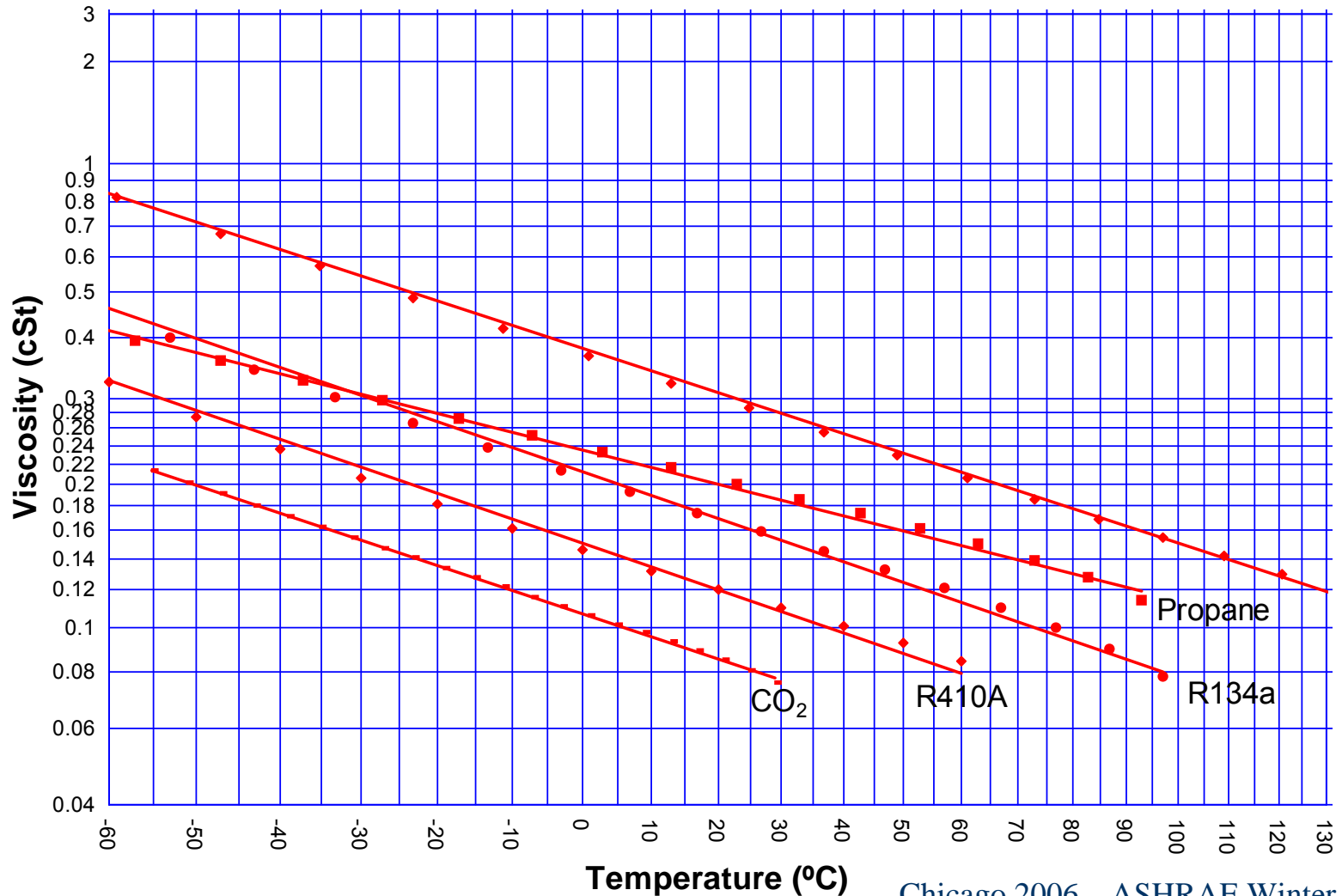
$$\psi = 1.244\ 067$$

$$\text{Log}_e \left( \text{Log}_e \left( \nu + 0.7 + e^{-\nu} K_0(\nu + 1.244067) \right) \right) = A - B * \text{Log}_e(T)$$

$$y = mx + b$$



# New Viscosity Chart



# ASTM Blending Rule

- Volume based blending rule

$$\text{Log}_{10} \left( \text{Log}_{10} \left( v_{mix} + 0.7 + f_{ASTM} (v_{mix}) \right) \right) = \sum V_i * \left( A_i - B_i * \text{Log}_{10} (T) \right)$$

$$V_i = \frac{\text{Vol}_i}{\text{Vol}_{total}} \qquad \text{Vol}_{total} = \sum \text{Vol}_i$$

$$y_{mix} = \sum V_i * \left( A_i - B_i * \text{Log}_{10} (T) \right)$$

For lubricants only, accuracy is generally better than  $\pm 15\%$

# Estimation depends on prior knowledge

- Assume density of fluids and their mixture are equal

Volume fraction = Mass Fraction

$$V_i = w_i$$

$$y_{mix} = \sum w_i * \left( A_i - B_i * \text{Log}_e (T) \right)$$

For refrigerant/lubricant mixtures, accuracy is generally better than  $\pm 20\%$

# Estimation depends on prior knowledge

- Recognize that the density of individual fluids are different, but they mix ideally

$$V_i = \frac{Vol_i}{Vol_{total}} = w_i \frac{\rho_{mix}(T)}{\rho_i(T)} \quad \rho_{mix}(T) = \left( \sum \frac{w_i}{\rho_i(T)} \right)^{-1}$$

$$y_{mix} = \sum_i w_i \frac{\rho_{mix}(T)}{\rho_i(T)} * (A_i - B_i * \text{Log}_e(T))$$

For refrigerant/lubricant mixtures, accuracy is generally better than  $\pm 15\%$

# Estimation depends on prior knowledge

- If measured data is available...

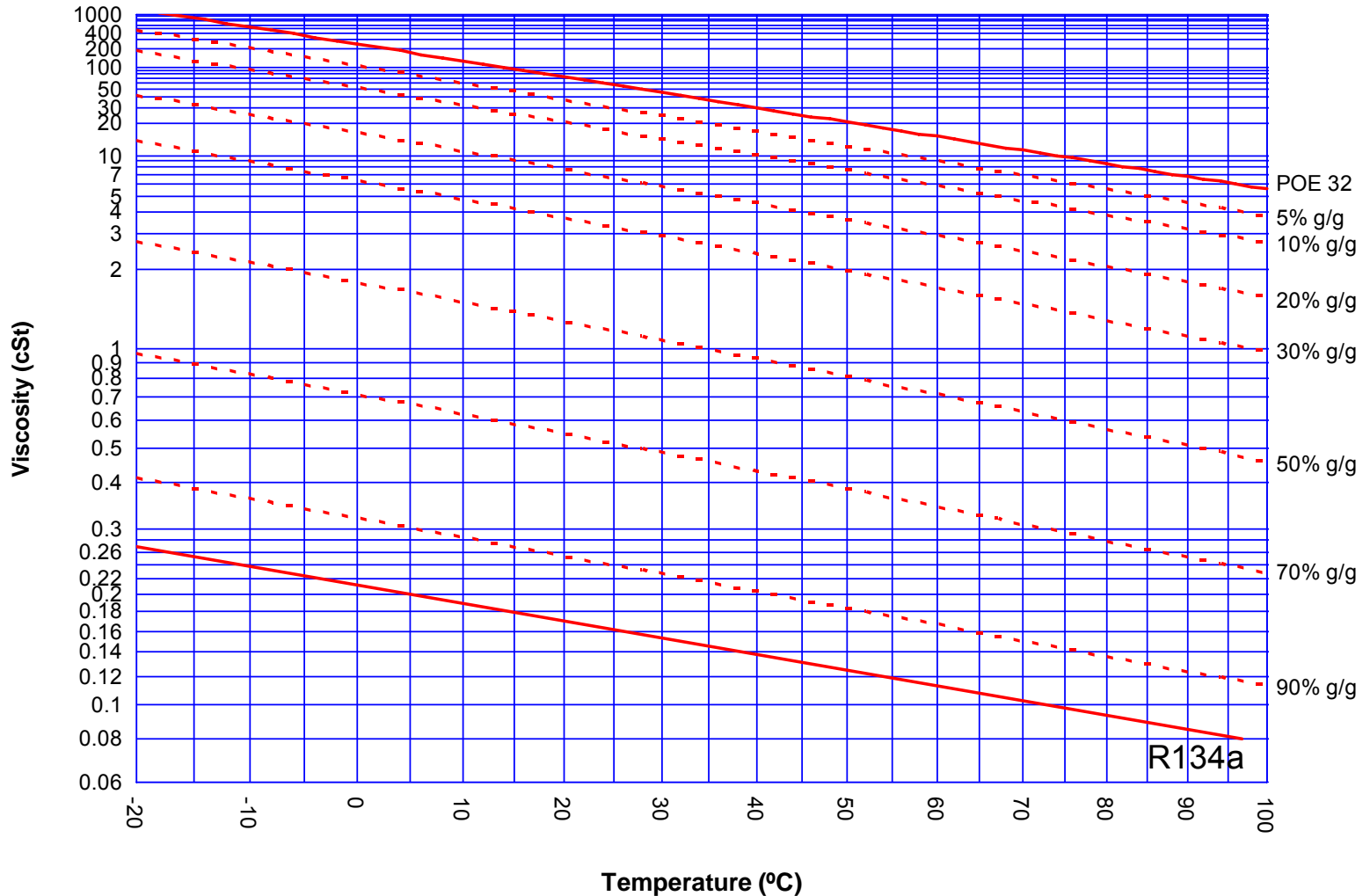
$$y_{mix} = \sum_i \left( \left( 1 + \sum_j \varphi_{ij} w_i w_j \right) * w_i \frac{\rho_{mix}(T)}{\rho_i(T)} * (A_i - B_i * \text{Log}_e(T)) \right)$$

$$\varphi_{ij} = \varphi_{ji} \quad ; \quad \varphi_{ii} = \varphi_{jj} = 0$$

$$\varphi_{ij} = f(T, w)$$

Accuracy is greatly enhanced  $\sim \pm 2\%$

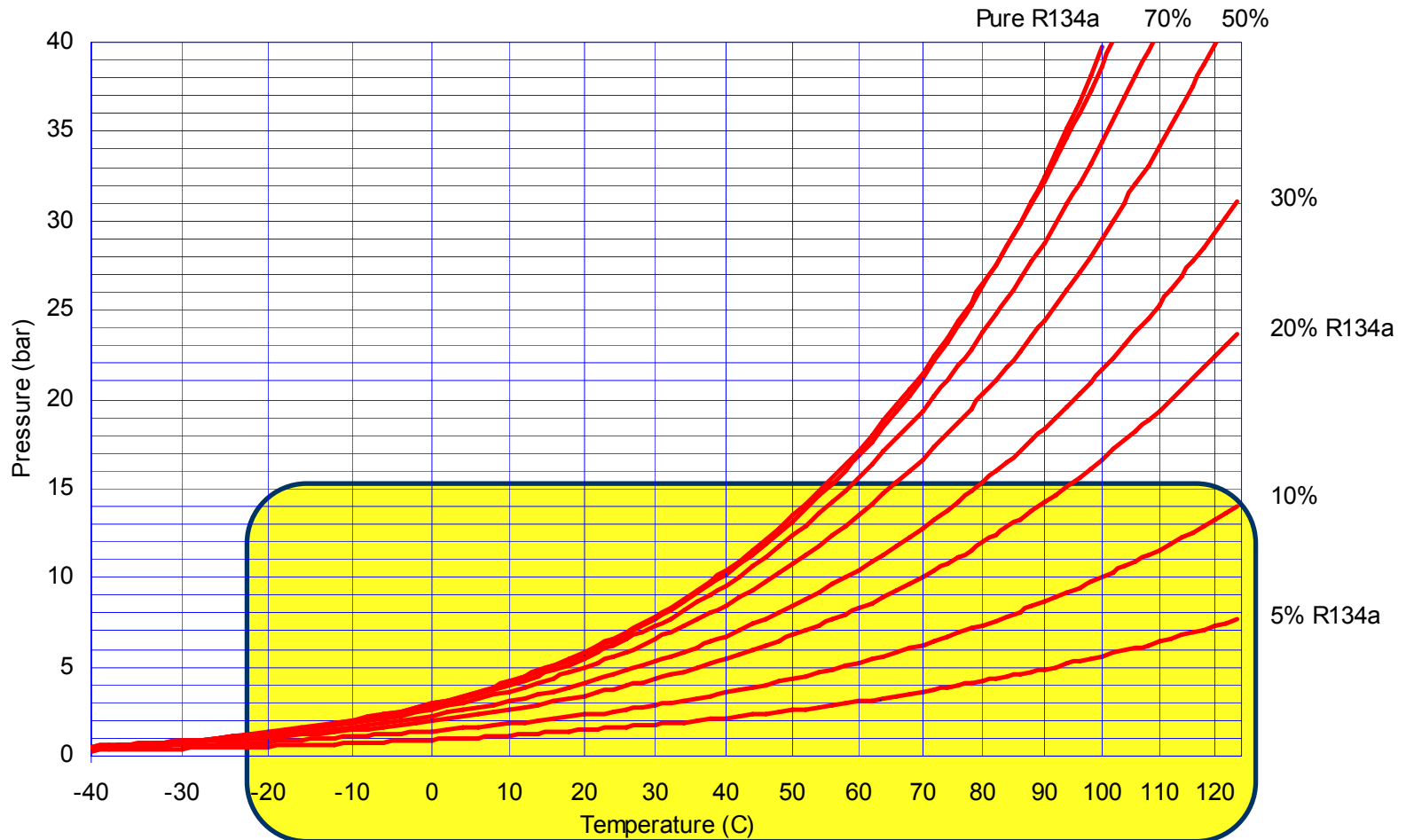
# Viscosity Prediction – POE + R134a



# Basic Solubility Measurement

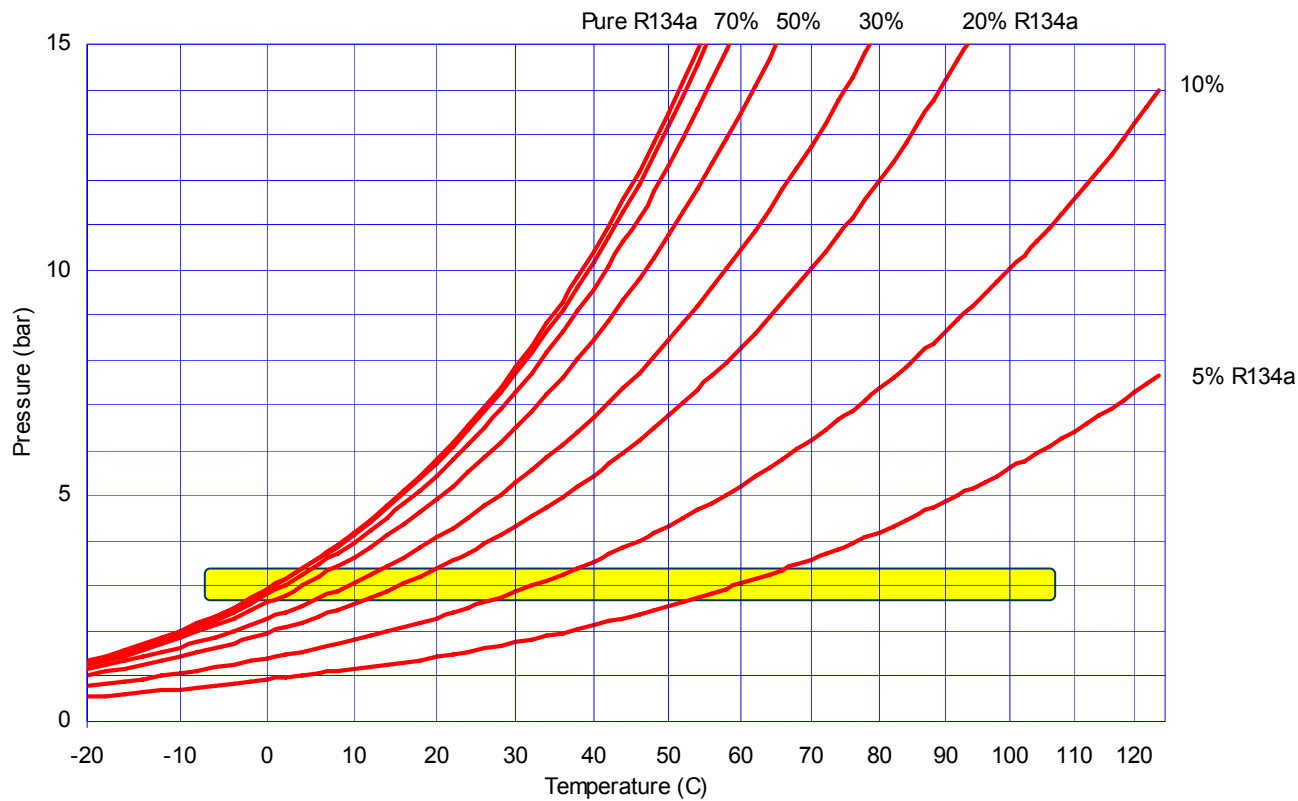


# Results of Solubility Measurements





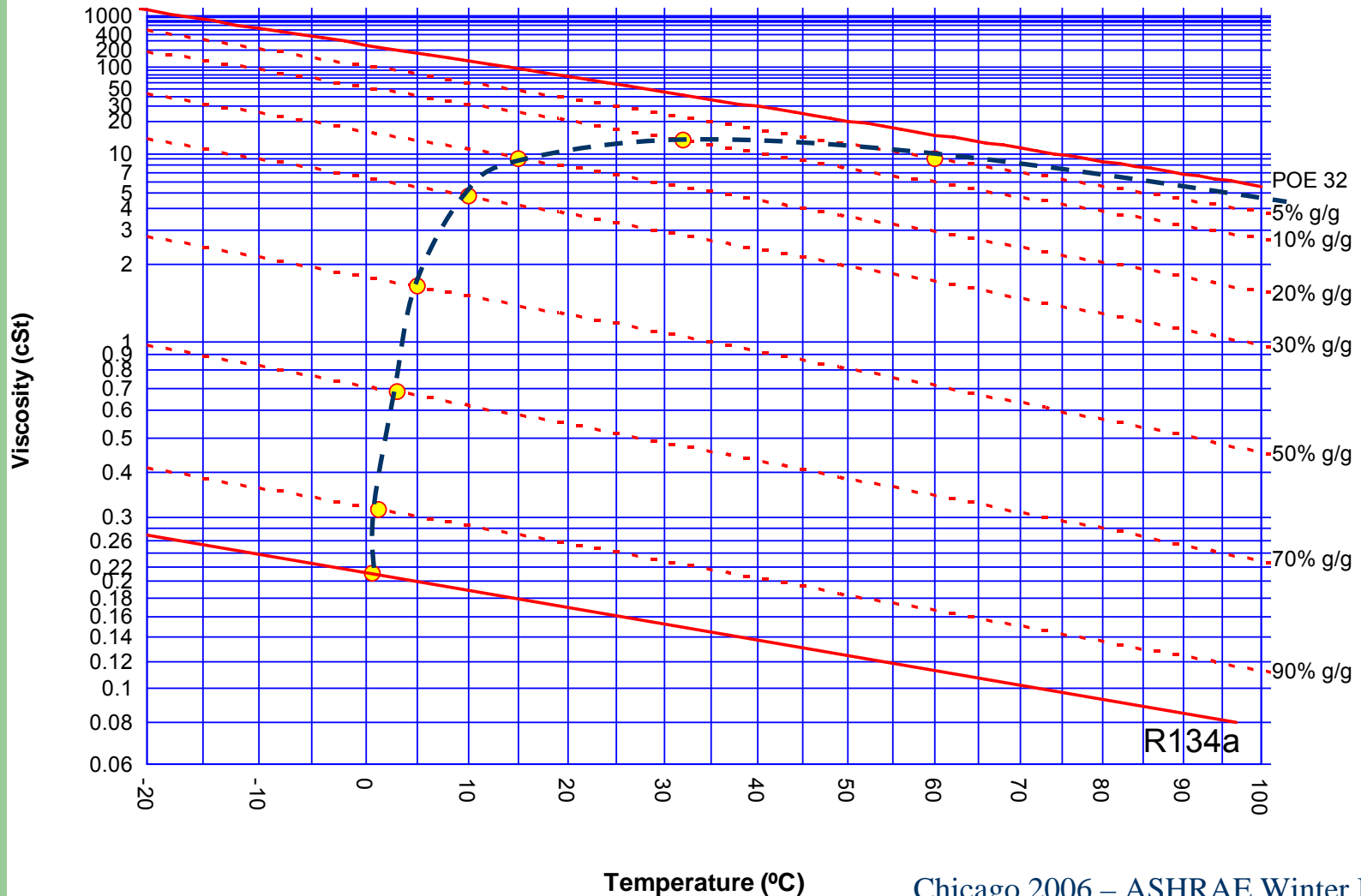
# Results of Solubility Measurements



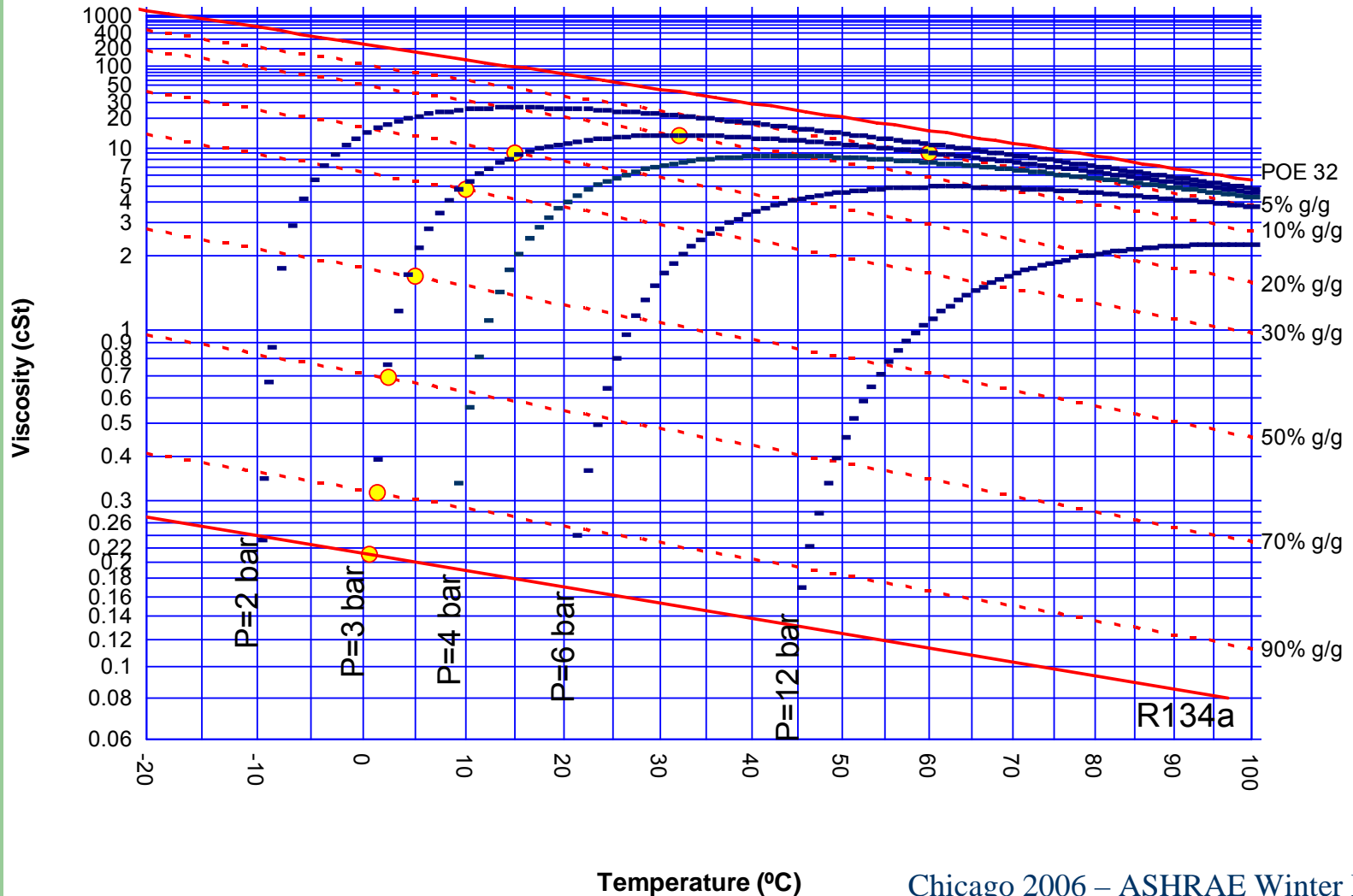
Pressure = 3 bar

Conc. % R134a	Temp. °C
5%	60
10%	32
20%	15
30%	10
50%	5
70%	2.5
90%	1.3
100%	0.67

# Constant Pressure Curves

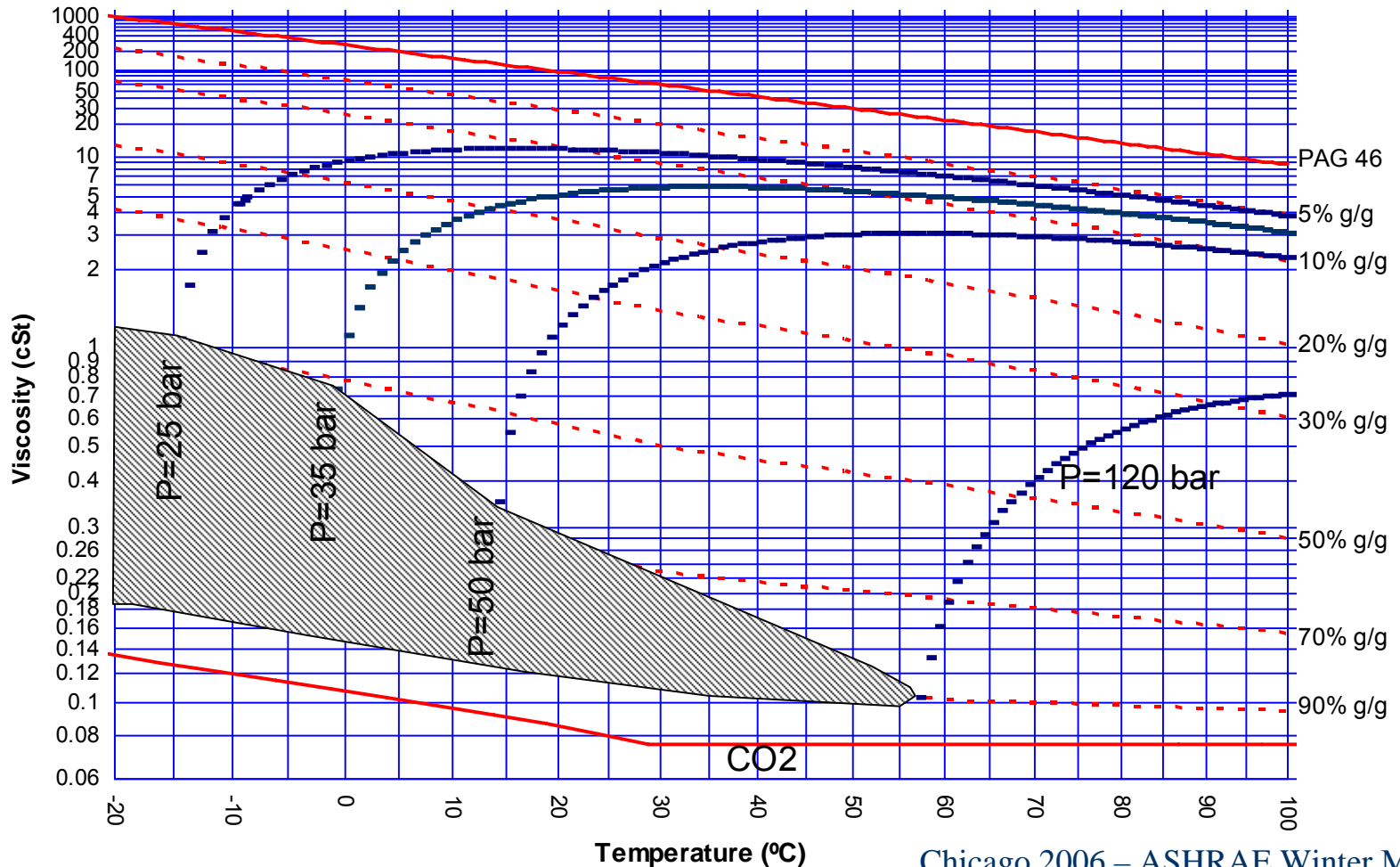


# Daniel Chart – POE32 / R-134a



# Real Motivation - CO<sub>2</sub>

**Estimated viscosity does not represent experimental measurements!**



# Conclusions

- Refrigerant – lubricant mixture viscosities can be estimated
- HFCs and hydrocarbons behave well on the proposed chart
- CO<sub>2</sub> does not mix ideally.
  - A new mixing rule is under development.
  - Experimental data is required!
    - Pressures must be over 35 bar (500 psia) to capture “in operation” conditions for air conditioning